

## Continuous Measurements of NO<sub>y</sub> and NO<sub>y</sub> Components during the Southeast Atmospheric Study, June 1-July 15, 2013

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Atmospheric dry deposition of oxidized nitrogen species (collectively known as NO<sub>y</sub>) is a complex function of chemistry and physics, including physical form (gas or aerosol), diffusivity, solubility and reactivity of individual species. Thus, information on individual components of NO<sub>y</sub> is needed to develop and/or constrain deposition estimates. This presentation will describe observations of NO<sub>y</sub> and its major components at a rural site in the southeastern US during the summer of 2013. Hourly and 5-minute average measurements of NO<sub>y</sub>, NO, NO<sub>2</sub>, HNO<sub>3</sub>, fine particulate nitrate (pNO<sub>3</sub>), total peroxy nitrates (tPANs) and total alkyl nitrates (tANs) were performed at the SEARCH Centreville, AL site during the multi-agency SAS/SOAS/SENEX campaign (June 1-July 15, 2013). All components were quantified via NO-O<sub>3</sub> chemiluminescence: NO directly, NO<sub>y</sub> via conversion to NO on 350C Mo, NO<sub>2</sub> via blue LED photolysis, HNO<sub>3</sub> via denuder difference, pNO<sub>3</sub> via denuder-filter difference, tPANs and tANs via thermal-photolytic conversion at 160C and 380C, respectively. Results show that the sum of components accounts for 98 +/- 11% of NO<sub>y</sub> (mean = 1250 +/- 810 parts per trillion) over the course of the study and that NO<sub>x</sub> (NO<sub>2</sub> + NO) and NO<sub>z</sub> (others) each account for almost exactly 50% of NO<sub>y</sub>. HNO<sub>3</sub>, tPANs, tANs and pNO<sub>3</sub> account for 12%, 10%, 21% and 4% of NO<sub>y</sub>, respectively. Each of the measured components has a more or less unique diurnal pattern. NO<sub>x</sub> accounts for >75% of NO<sub>y</sub> during nighttime hours and is exclusively NO<sub>2</sub> overnight, while HNO<sub>3</sub>, NO<sub>2</sub>, tPANs and tANs each account for roughly 25% of NO<sub>y</sub> during midday. HNO<sub>3</sub> concentrations generally peak around 1600 local standard time (LST), while tPANs and tANs generally peak around 0900-1200 LST. pNO<sub>3</sub> concentrations peak within +/- 2 hours of sunrise and are effectively zero from 1000-1600 LST. Inspection of time series data provides interesting clues to the physical-chemical characteristics of tPANs and tANs (recognizing, of course, that these are classes of compounds rather than individual compounds). For example, tPAN and tAN concentrations are very slightly affected by rainfall events, whereas HNO<sub>3</sub> and pNO<sub>3</sub> concentrations are rapidly reduced to zero or near zero. Similarly, nighttime concentrations of tPANs and tANs decline much more slowly than HNO<sub>3</sub>. These observations suggest that tPANs and tANs have substantially different (slower) wet and dry removal rates than HNO<sub>3</sub>, which, in turn, may provide useful information for estimating NO<sub>y</sub> deposition.

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